

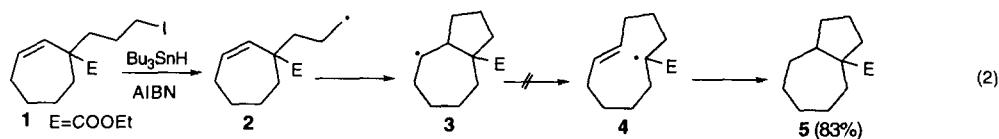
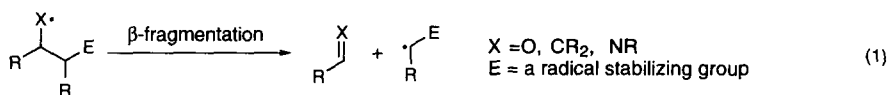
The Effect of α -Alkoxy Group in Radical-Mediated β -Fragmentation Reactions

Sungak Kim,* Kwan Hee Kim, and Jin Rai Cho

Department of Chemistry, Korea Advanced Institute of Science and Technology, Taejeon 305-701, Korea

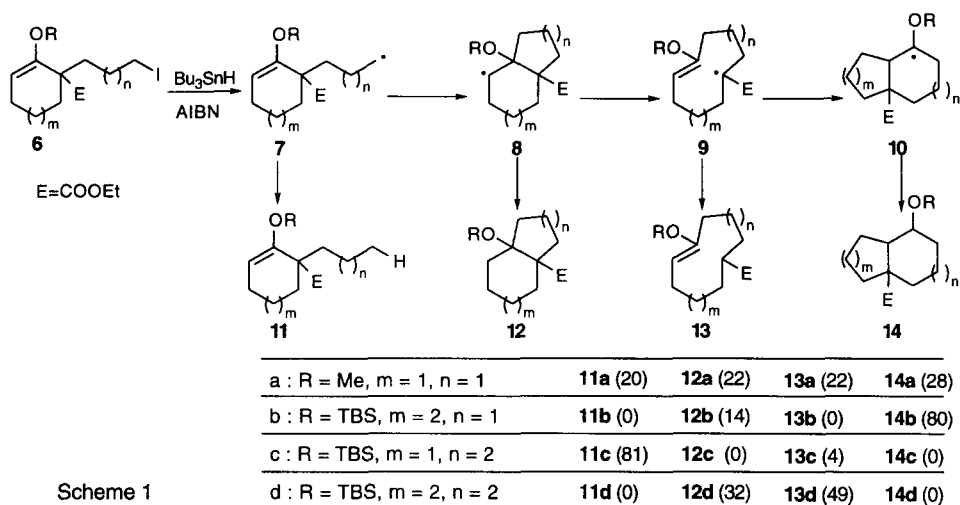
Summary: β -Fragmentation reactions of alkyl and aminyl radicals were greatly facilitated by the presence of α -alkoxy groups. © 1997 Elsevier Science Ltd.

β -Fragmentations in radical reactions are unique and versatile, and they are quite different from β -fragmentations in ionic reactions.¹ The driving forces for β -fragmentations in radical reactions are often (i) relief of ring strain,² (ii) cleavage of carbon-heteroatom bonds,³ (iii) formation of π bonds along with cleavage of C-C bonds (eq 1).⁴ Among three different types of driving forces for β -fragmentations, we were interested in type iii, where cycloalkyloxy radicals normally undergo facile β -fragmentations because of strong π -bond energy of C=O bonds. β -Fragmentations of alkyl and aminyl radicals do not normally occur, although their β -fragmentations would produce more stable radicals. Thus, β -fragmentations involving the formation of C=C and C=N bonds are very rare.⁵ We wish to report that the presence of α -alkoxy groups facilitated β -fragmentations of alkyl and aminyl radicals significantly.



We initially examined the possibility of β -fragmentations of alkyl radicals involving the formation of C=C bonds along with the generation of stable radicals (eq 2). The reaction of iodide **1** with $\text{Bu}_3\text{SnH}/\text{AIBN}$ in refluxing benzene under a high dilution for 3 h afforded only **5** in 83% yield. Apparently, β -fragmentation of the alkyl radical **3** into **4** did not take place.

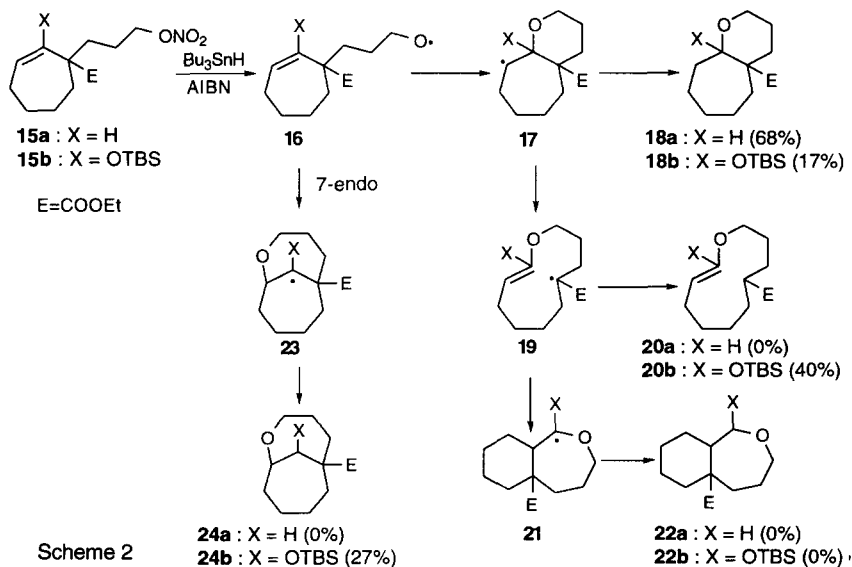
In order to see the effect of α -alkoxy group in β -fragmentation, we prepared substrate **6** by routine operations. As shown in Scheme 1, when **6** is treated with $\text{Bu}_3\text{SnH}/\text{AIBN}$, four products are expected to be formed, where the latter two products (**13** and **14**) would be produced via β -fragmentation of **8**. Furthermore, the ratio of **13** and **14** would depend on the size of the ring (m,n). As predicted, the radical reaction of **6a** with $\text{Bu}_3\text{SnH}/\text{AIBN}$ in refluxing benzene under a high dilution afforded a mixture of four products roughly in an equal ratio.⁶ A much better result was obtained with **6b**, where **14b** was isolated in 80% yield along with **12b** (14%). Since we were unable to observe any β -fragmentation products with **1**, this result clearly



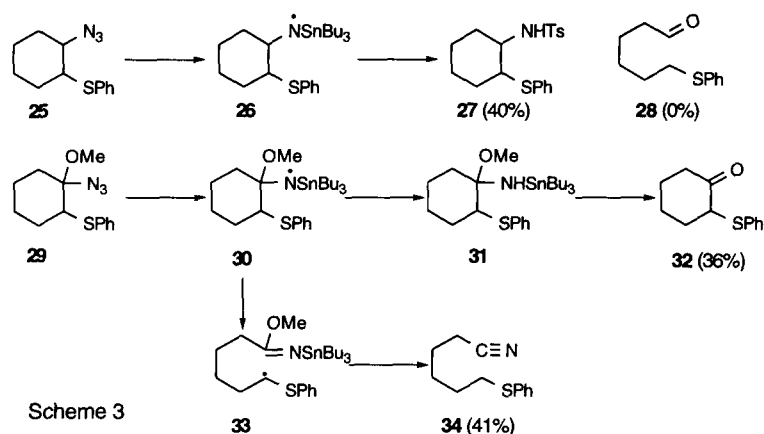
Scheme 1

demonstrates the importance of α -silyloxy group in β -fragmentation of alkyl radical **8b**. In the case of **6c**, the direct reduction product was obtained in 81% yield along with a small amount of β -fragmentation product **13c** (4%). Furthermore, when **6d** was subjected to the similar conditions, the β -fragmentation product **13d** (49%) was favored over the cyclized product **12d** (32%). Although the ratio of β -fragmentation relative to direct quenching and cyclization depends very much on the size of the ring, it is evident that the presence of α -alkoxy group certainly facilitates the β -fragmentation of alkyl radicals.

In order to examine the scope and limitations of the effect of α -alkoxy groups, we studied the radical reaction of **15a**,⁷ in which the alkoxy group was generated by intramolecular addition of an alkoxy radical to an alkenyl group. Although an α -alkoxy group was available, β -fragmentation in **17a** (X=H) did

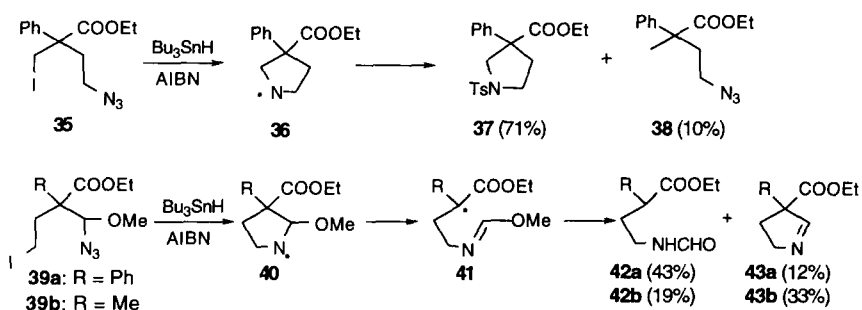


Scheme 2

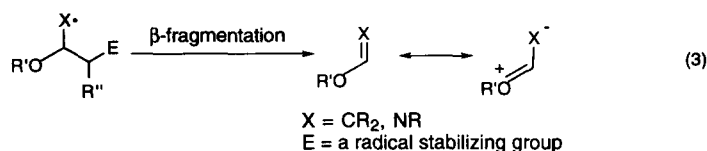


not occur, yielding only cyclized product **18a** in 68% yield. However, the presence of another α -alkoxy group facilitated the β -fragmentation of an alkyl radical. Thus, when **15b** was subjected to the similar conditions,⁶ β -fragmentation product **20b** was obtained in 40% yield along with **18b** (17%). In addition, somewhat surprisingly, 7-*endo* addition product **24b** was also isolated in 27% yield.⁸

Our attention was next turned to the question of whether α -alkoxy groups would also facilitate β -fragmentations involving the formation of C=N bonds. As shown in Scheme 3, azido groups were utilized to generate aminyl radicals.⁹ Radical reaction of **25** with $\text{Bu}_3\text{SnH/AIBN}$ in refluxing benzene⁶ afforded direct reduction product **27** (40%) after tosylation along with diphenyl disulfide (50%). There was no indication of the presence of β -fragmentation product **28**. However, when the reaction was carried out with **29** bearing an α -methoxy group under the similar conditions, a mixture of **32** and **34** was isolated in 36% and 41% yield, respectively. Apparently, **34** was produced via β -fragmentation of aminyl radical **30** and subsequent thermal elimination of tributyltin methoxide.



We next examined the effect of α -alkoxy group in the cyclic aminyl radicals as shown in Scheme 4. As we previously reported the radical cyclization of alkyl azides,^{9b} when azide **35** was treated with $\text{Bu}_3\text{SnH/AIBN}$ under a high dilution, the cyclized product **37** was isolated in 71% yield along with direct reduction product **38** (10%). However, in **40** bearing α -alkoxy group, β -fragmentation occurred to some extent, depending on the stability of the resulting radicals. When **39a** was treated with $\text{Bu}_3\text{SnH/AIBN}$,⁶ β -fragmentation product **42a** was obtained in 43% yield along with quenching product **43a** (12%) after thermal elimination of methanol, indicative of the importance of α -alkoxy group in the β -fragmentations



involving the formation of C=N bonds. A similar result was also obtained with **39b**, although a less amount of β -fragmentation product **42b** was isolated.

The experimental results obtained in this study clearly indicate that the β -fragmentations of alkyl and aminyl radicals are greatly facilitated in the presence of α -alkoxy groups. The reason for this observation is not clear at the present. We assume that the resonance stabilization of β -fragmentation products due to the presence of α -alkoxy group would facilitate the β -fragmentations of alkyl and aminyl radicals (eq 3).

Acknowledgement. This work was financially supported by the Organic Chemistry Research Center (OCRC) and KAIST.

References and Notes

- For reviews, see: (a) Curran, D. P. *Synthesis* **1988**, 489. (b) Jasperse, C. P.; Curran, D. P.; Fevig, T. L. *Chem. Rev.* **1991**, *91*, 1237. (c) Dowd, P.; Zhang, W. *Chem. Rev.* **1993**, *93*, 2091.
- For selected examples, see: (a) Ayril-Kaloustian, S.; Agosta, W. C. *J. Am. Chem. Soc.* **1980**, *102*, 314. (b) Maeda, Y.; Ingold, K. U. *J. Am. Chem. Soc.* **1980**, *102*, 328. (c) Kim, S.; Lee, S.; Koh, J. S. *J. Am. Chem. Soc.* **1991**, *113*, 5106. (d) Kim, S.; Kee, I. S.; Lee, S. *J. Am. Chem. Soc.* **1991**, *113*, 9882. (e) Crimmins, M. T.; Dudek, C. M.; Cheung, W.-H. *Tetrahedron Lett.* **1992**, *33*, 181. (f) Newcomb, M. *Tetrahedron* **1993**, *49*, 1151.
- (a) Keck, G. E.; Yates, J. B. *J. Am. Chem. Soc.* **1982**, *104*, 5829. (b) Barton, D. H. R.; Crich, D. *Tetrahedron Lett.* **1984**, *25*, 2787. (c) Keck, G. E.; Byers, J. H. *J. Org. Chem.* **1985**, *50*, 5444. (d) Baldwin, J. E.; Adlington, R. M.; Robertson, J. *J. Chem. Soc., Chem. Commun.* **1988**, 1404.
- (a) Beckwith, A. L. J.; Kazlauskas, R.; Syner-Lyons, M. R. *J. Org. Chem.* **1983**, *48*, 4718. (b) Dowd, P.; Choi, S.-C. *J. Am. Chem. Soc.* **1987**, *109*, 6548. (c) Sugimoto, H.; Yamada, S. *Tetrahedron* **1987**, *43*, 3371. (d) O'Dell, D. E.; Loper, J. T.; Macdonald, T. L. *J. Org. Chem.* **1988**, *53*, 5225. (e) Beckwith, A. L. J.; O'Shea, D. M.; Westwood, S. W. *J. Am. Chem. Soc.* **1988**, *110*, 2565. (f) Murphy, J. A.; Patterson, C. W.; Wooster, N. *Tetrahedron Lett.* **1988**, *29*, 955. (g) Walton, W.; Fraser-Reid, B. *J. Am. Chem. Soc.* **1991**, *113*, 5791. (h) Rawal, V. H.; Zhong, H. M. *Tetrahedron Lett.* **1993**, *34*, 5197. (i) Callier, A.-C.; Quiclet-Sire, B.; Zard, S. Z. *Tetrahedron Lett.* **1994**, *35*, 6109.
- (a) Bowman, W. R.; Clark, D. N.; Marmon, R. J. *Tetrahedron Lett.* **1991**, *32*, 6441. (b) Bowman, W. R.; Clark, D. N.; Marmon, R. J. *Tetrahedron*, **1994**, *50*, 1275. (c) Ryu, I.; Muraoka, H.; Kambe, N.; Komatsu, M.; Sonoda, N. *J. Org. Chem.* **1996**, *61*, 6396.
- The reaction was carried out by the addition of a 0.05M benzene solution of Bu₃SnH (1.1 equiv)/AIBN (0.1 equiv) by a syringe pump for 3 h to a 0.05M refluxing benzene solution of iodide **6**. The remaining reactions were carried out under the similar conditions.
- (a) Fraser-Reid, B.; Vite, G. D.; Yeung, B.-W. A.; Tsang, R. *Tetrahedron Lett.* **1988**, *29*, 1645. (b) Vite, G. D.; Fraser-Reid, B. *Synth. Commun.* **1988**, *18*, 1339. (c) Lopez, J. C.; Alonso, R.; Fraser-Reid, B. *J. Am. Chem. Soc.* **1989**, *111*, 6471.
- (a) Fossey, J.; Lefort, D.; Sorba, J. *Free Radicals in Organic Chemistry*, p 248, John Wiley & Sons, 1995. (b) Bachi, M. D.; Hoornaert, C. *Tetrahedron Lett.* **1982**, *23*, 2505. (c) Bachi, M. D.; Frolow, F.; Hoornaert, C. *J. Org. Chem.* **1983**, *48*, 1841. (d) Colombo, L.; Giacomo, M. D.; Scolastico, C.; Manzoni, L.; Belvisi, L.; Moteni, V. *Tetrahedron Lett.* **1995**, *36*, 625.
- (a) Kim, S.; Joe, G. H.; Do, J. Y. *J. Am. Chem. Soc.* **1993**, *115*, 3328. (b) Kim, S.; Joe, G. H.; Do, J. Y. *J. Am. Chem. Soc.* **1994**, *116*, 5521.

(Received in Japan 4 March 1997; revised 14 April 1997; accepted 21 April 1997)